ALEXANDRIA science exchange journal

VOL. 37

JULY-SEPTEMBER

2016

Evaluation of Water Quality and Heavy Metal Indices of Some Water Resources at Kafr El-Dawar Region, Egypt

Aggag A.M.¹

ABSTRACT

Kafer El-Dawr is a major industrial and municipality city at the western area of Nile Delta, northern Egypt. Many factories are founded in Kafr El-Dawar region that discharges many pollutant elements. Assessment of Water quality is an important issue to know whether it is safe or not for irrigation. So, twenty-five water samples were collected from some water sources established in Kafer El-Dawr region. These sources are Yarn and Fabric (YD), Kafer El-Dawr Defshu (KDD), Dabora Abu Qir (DAD), Dabora Defshu Canal (DC) and Abu Qir (AD). Whisker box plot-median indicated that the heavy metals in the waters of these drains can be classified according to their concentration homogeneity into (a) wide spreadheterogeneous included: Cr, Co and Cu. (b) Moderate spread - moderately homogeneous included: Pb, Ni and Zn, and (c) narrow spread - homogeneous included: Fe, Cd, Li and Mn. The correlation study classified the heavy metals into three groups: The positively highly correlation between Pb and Cd, Co and Cr concluded that the water resources have the same pollution source. Contrary, the negatively highly correlated between (Pb_Mn), (Mn_Cu) and (Co_Li) might be a tool to assume that these heavy metals originated from different pollution resources.

Water quality for irrigation was evaluated by water quality index (WQI), heavy metals pollution in short and long-term use by conventional scale, heavy metals contamination index (CI) and metal index (MI). According to EPA the concentrations of heavy metals was generally safe except cadmium for short-term use. For long-term use, Co occupied the polluted class in all water resources and Mn in water resource (KDD). The risky pollution was found in Cd and Cr in all resources as well as Mn in (DAD) and (AD) resources.

Water quality index (WQI) showed that these water resources are good for irrigation utilization. Metal index values cleared that drains or canal are seriously threatened with metal pollution for irrigation usage (MI>1). Only KDD drain has no heavy metal problems to use in irrigation. The contamination index (CI) showed that the water resources (YD), (DAD) and (AD) had negative values of -5.67, -7.26 and -7.49, respectively. This is indicated that these water resources are safe to use in short-term run. Contrary, all studied water resources were highly contaminated and cannot be use at the long run.

Keywords: Heavy metals, Kafr El-Dawar, Water quality index, contamination index, Metal index

INTRODUCTION

Water pollution of natural water bodies like lakes, rivers, streams, oceans, and groundwater is due to inflow or deposition of pollutants directly or indirectly into water systems. Pollution very often is caused by human activities (Mwegoha and Kihampa, 2010). The use of water for different purposes such as drinking, irrigation, domestic and industrial, mainly depends on its intrinsic quality. So, It is necessary to examine quality of water resources available in the region (Mohrir, 2002). Owing to industrial and agricultural activities large amounts of untreated urban, municipal and industrial wastewater are discharging into the canals or agricultural drains which become an easy dump sites for wastes containing toxic metals (Karbassi and Bayati, 2005 and Goher *et al.*, 2014).

Heavy metals are regard as serious pollutant of aquatic ecosystem because of their environmental persistence and toxicity effects on living organisms (Khalil *et al.*, 2007). The increased load of heavy metals in the aquatic ecosystems have severely disrupted water quality which threatend aquatic organisms and human health (Sasmaz *et al.*, 2008 and Elshemy and Meon, 2011).

Heavy metal water pollution has been studied by standard tables of pollution (EPA, 2004) or by applying contamination indices, such as contamination index (CI), water quality index (WQI) and Metal Index (MI) as reported by Brraich and Jangu (2015) and Manoj *et al.*, (2012). The water quality index (WQI) is a single number that expresses water quality by aggregating the measurements of water quality parameters (Lumb et al., 2011). A Wisker boxplot can give information regarding the shape, variability, and center (or median)

Ass. Prof., Nat. Res. & Agric. Eng. Dept., Fac. Agric.,

Damanhour Univ., Egypt (www.damanhour.edu.eg)

Email: aaggag5@dmu.edu.eg

Received July 17, 2016, Accepted August 2, 2016

of a statistical data set. It is particularly useful for displaying skewed data (Rumsey, 2016). This boxplot can graphically present heavy metal distribution pattern (Sany *et al*, 2013).

At Kafr El-Dawar region, west Nile Delta, the sources of pollutants are due to the largest industrial zones in the region which include the activities of Albayda Dyers Co., Albayda Tinning and Chemicals Co., Egyptian Textile and Spinning which caused dramatic changes in its water quality by discharging wastewater.

This study aims to: (a) assess the water quality index approach for some water resources in Kafr El-Dawar region, (b) statistical analysis approaches of the water quality data such as correlation analysis, (c) assess heavy metals pollution risk, and (d) improve capability of water environmental monitoring and supervision.

MATERIALS AND METHODS

Kafer El-Dawar region is а major industrial and municipality city in the western area of Nile Delta, northern Egypt. Twenty Five water samples were collected from some water resources, Kafer El-Dawr region: Yarn and fabric Drain (YD), Kafer Dawr Defshu Drain (KDD), Dabora Abu Qir Drain (DAD), Dabora Defshu Canal (DC) and Abu Qir Drain (AD). Details of water sampling locations along with their longitude and latitude are presented in Fig. 1 and Table 1. Several heavy metals were determined in the water recourses.

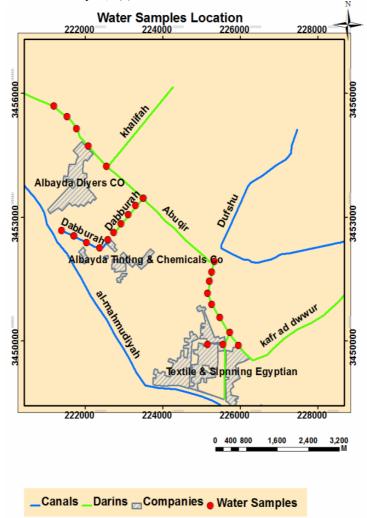


Fig 1. Location of water samples resources in Kafr El-Dawar region

Water reso	ource	Coordina	tes (UTM)	Water re	esource	Coordina	ates (UTM)
Name	Symbol	Е	N	Name	Symbol	E	N
		225159	3449988			222627	3452223
Yarn and fabric		225553	3449908	Dabora		222920	3452831
Drain	YD	225726	3450183	Defshu	DC	223110	3453057
Jiaill		225949	3449891	Canal		223299	3453281
		225467	3450562	_		223488	3453453
		225467	3454498			222544	3454497
CAFR El-	KDD	225261	3450889			221475	3455484
DAWR Defshu		225158	3451147	- Abu Qir	AD	221527	3455432
Drain		225261	3451663	- Drain		221767	3455140
		2253330	3451922	_		222077	3454727
Dahama Ahu Oin		222573	3452200	Dabora Abu	DAD	222026	3452386
Dabora Abu Qir	DAD	221389	3452696	Qir Drain	DAD	222370	3452249
Drain		221699	3452541				

 Table 1. Water samples coordinates

Laboratory analysis:

Water sampling technique was carried out according to the methods outlined in APHA (2005). The area samples were kept in 25 polyethylene bottles in ice box for analysis in the laboratory. The methods of analyses were carried out according to Page *et al.* (1982) and Ayers and Westcot (1994).

Water electrical conductivity (EC) and pH value were measured in situ, using YSI model. Chloride was measured using Mohr's method and Calcium and magnesium were determined by direct titration using versinate method (Na₂ EDTA), Na⁺ and K⁺ were measured using flame photometer Model "Corning 400". Sodium Adsorption Ratio (SAR) was calculated. Heavy metals (Cd⁺², Cu⁺², Fe⁺², Mn⁺², Ni⁺², Pb⁺², Cr⁺², Co⁺², Li⁺² and Zn⁺²) were measured using atomic absorption Spectrophotometer (Thermo model iCE 3300).

Water quality index:

Water quality index (WQI) is defined as a technique of rating that provides the composite influence of individual water quality parameter on the overall quality of water (Al-Mohammed and Mutasher, 2013). WQI has been calculated to evaluate the suitability of water resources for irrigation using the Weighted arithmetic water quality index method, which classifies the water quality according to the degree of purity by using the most commonly measured water variables. The method of calculation has been widely used by many scientists (Tyagi1 *et al.*, 2013; Chowdhury *et al.*, 2012; Balan *et al.*, 2012). The mathematical formula of this WQI method is given by:

WQI =
$$\sum_{i=1}^{n} Q_i W_i / \sum_{i=1}^{n} W_i$$

where Q_i is the sub quality index of *i*th parameter (or Q_i is the quality rating scale of each parameter). W = weight unit of each parameter, n = number of parameters. The Q_i can be calculated as:

$$Q_i = [(V_i - V_0)/(S_i - V_0)]$$

 V_i = measured value of *i*th parameter, S_i =standard permissible value of *i*th parameter, Vo =ideal value of *i*th parameter in pure water, V_0 = zero for all parameters except for pH =7.0 (Tripaty and Sahu, 2005).

The contribution or importance to water quality of each indicator is usually different, and can be indicated by weighting coefficient. The calculation of weights assigned to each indicator is as follows (Lumb *et al.*, 2011):

- 1- The sum squared deviation from the mean was obtained for each observation,
- 2- This amount was summed up for all observations for a specific indictor,
- 3- Obtaining the total sum squared deviation from the mean for all indictors,
- 4- The weight was obtained by dividing step 2 by step 3 and
- 5- The sum of all weights was normalized to 100%.

WQI has been classified into 5 classes; excellent, good, poor, very poor and unfit when the value of the index lies between 0-25, 26-50, 51-75, 76-100 and >100, respectively (Table 2).

Two different quality indices are used to determine the metal contamination of water source.

uiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiiii	water quality match met	nou
WQI value	Rating of Water Quality	Grading
0-25	Excellent	А
26-50	Good	В
51-75	Poor	С
76-100	Very Poor	D
Above 100	Unsuitable (unfit)	Е

 Table 2. Water quality rating as per weight arithmetic water quality index method

(i) Contamination Index (CI):

The contamination index (CI) summarizes the combined effects of several quality parameters considered harmful to irrigation (Hakanson 1980). This index is calculated from the formula:

$$CI = \sum_{i=1}^{n} C_{fi}$$

Where:

$$C_{\text{fi}} = \frac{C_{Ai}}{C_{Ni}} - 1$$

Where C_{fi} , C_{Ai} and C_{Ni} represent contamination factor, analytical value and upper permissible concentration of the *i*th component, repectively (N denotes the "normative value"). Interpretative scale of contamination factor (C_{fi}) is illustrated in Table (3) which conducted to describe individually the pollution potentialty of each heavy metal.

Table 3. Interpretative scale of contamination factor (C_{fi})

N	C _{fi} value	C _{fi} Category
1	-0.5 - 0.0	Safe
2	0.0 - 9.0	Risky
3	> 9.0	Polluted

(ii) Metal index (MI): is based on a total trend evaluation of the present status. The higher the concentration of a metal compared to its respective maximum allowable concentration (MAC) value, the worse the quality of the water. MI value >1 is a threshold of warning (Bakan *et al.*, 2010). According to (Tamasi and Cini, 2004), the MI is calculated by using the following formula:

$$MI = \sum_{i=1}^{n} \frac{C_i}{(MAC)_i}$$

Where: C_i: the concentration of each element, MAC: maximum allowable concentration.

Statistical analysis:

Data were analyzed for descriptive statistics to measure homogeneity of data that may indicate the HM resources. Whisker box Plot was drawn to represent graphically the HM distribution patterns. Data also classified using the standard tables of EPA (2004). In addition, the correlation coefficient (r) between the measured parameters was examined.

RESULTS AND DISCUSSION

The minimum, maximum, mean, median values and standard deviations of the obtained results are presented in Table (4).

The values of EC showed a spatial difference and ranged between $830 - 2060 \ \mu \text{Sm}^{-1}$ and the pH values were in the alkaline side (7.44-8.66). There are high positive correlations between pH/Pb (r = 0.80), pH/Cd (r = 0.71) and high negative correlations between pH/EC (r = -0.64), pH/Ca²⁺ (r = -0.79), pH/Mg²⁺ (r = -0.71), pH/HCO₃⁻ (r = -0.62) and pH/Mn (r = -0.85) (Table 5). HCO_3^- and Cl concentrations varied between 195.2-518.5 and 92.3-284.0 mg/l, respectively. HCO₃and Chloride positively correlated with Mg⁺², Na⁺ and K⁺. Calcium and Magnesium values ranged between 60.0-160.0 and 19.2-66.0 mg/l, respectively. Sodium and potassium values ranged between 64.4-230.0 and 7.8-23.4 mg/l, respectively. The valued of sodium adsorption ratio (SAR) reported in this study ranged from 1.7 to 5.3.

The concentrations of Pb, Cd, Zn, Mn, Cr, Ni, Cu, Fe, Co and Li were in the ranges of (3.2-29.5), (1.0-10.0), (43.8-96.8), (7.6-63.5), (25.2-26.3), (13.1-14.0), (1.2-5.1), (3.0-98.0), (10.1-11.4) and (3.3-6.4) µg/l, respectively. According to Ibrahim and Omar (2013), the amount fluctuations of agricultural drainage water, sewage effluents and industrial wastes discharged into the drains or canal, are the main reasons for the difference in the concentration of heavy metal.

Water quality index:

Figure (4) showed that the values of WQI of some water resources in Kafr El-Dawar. The WQI score for irrigation water was computed using guidelines of FAO (1994). The results showed that WQI values ranged between 25.28 and 37.02 with respect to irrigation water according to the irrigation guidelines (Table 2). This study indicate that the water quality fluctuation of Kafr El-Dawar water resources could be classified as "good" water for irrigation utilizations.

Heavy Metals Content in Water Resources:

Heavy Metals water resources content was determined and listed in (Table, 5) and illustrated in Figure (3).

The determination of the heavy metals in water resources conducted to compose the different increasing concentration sequence of heavy metals (Figure, 3). The sequences showed that all studied water resources begin with Cu. The sequences of YD, DC and AD water resources ended by Zn as maximum concentration. Thus, the source of heavy metals contamination is the same in the three water resources. Whereas DAD and KDD water resources have Mn and Fe as maximum concentration.

Distribution Pattern of Heavy Metals in water resources

The descriptive statistics of heavy metals content in water resources (Table, 4) indicated that, there were great variations between the mean and median of heavy metals concentrations in waters. This heterogeneity of heavy metal concentration was confirmed by the high values of standard deviation (S.D.).

The distribution pattern was graphically presented by Whisker box plot –median (Fig 4: a, b, and c). This type of whisker plot was preferred because of heterogeneity of the data. This figure indicated that the population of heavy metals can be classified, according to their homogeneity of concentration, into three distribution patterns:

- Moderate spread moderately homogeneous subpopulation Pb , Ni and Zn
- Narrow spread homogeneous subpopulation; Fe ,Cd, Li and Mn.

Source of Heavy Metals Pollution:

The heavy metals (HM) intercorrelation matrix was calculated (Table, 6) to assess their associations. These associations conducted to classify the HM into three groups:

- Positively highly correlated HM: (Pb_Cd), (Pb_Co), (Pb_Cr), (Cd_Co), (Cd_Cu) (Zn_Fe), (Ni_Co) and (Ni_Fe)
- Negatively highly correlated HM: (Pb_Mn), (Mn_Cu) and (Co_Li)
- Intermediate cases (moderate positive and negative correlation): (Cr_Co), (Cd_Mn), (Pb_Li), (Zn_Ni), (Zn_Cu), (Cr_Li) and (Fe_Co).

The positively highly intercorrelated HM between Pb and Cd, Co and Cr led to conclude that the studied water resources have been submitted, by these four heavy metals to the same pollution source. Contrary, the negatively highly intercorrelated between (Pb_Mn), (Mn_Cu) and (Co_Li) might be a tool to assume that these heavy metals originated from different pollution resources. The intermediate cases of low positive and negative correlation coefficients indicated multiple pollution resources.

Table 4. Descriptive statistics of water parameters compared to guidelines used in WQI, CI and MI computations

Parameter	Min.	Max.	Mean	Median	S.D.	Guideline [*]
pН	7.44	8.66	8.142	8.4	0.54	8.5
EC (μ Sm ⁻¹)	830	2060	1506	1560	441.28	3000.0
Ca^{2+} (mg/l)	60	160	88.8	76	40.34	400
Mg^{2+} (mg/l)	19.2	66	42.48	45.6	17.01	60
Na ⁺ (mg/l)	64.4	230	188.6	211.6	69.91	919
K^{+} (mg/l)	7.8	23.4	16.38	15.6	5.78	2
HCO ₃ ⁻ (mg/l)	195.2	518.5	396.5	439.2	130.97	610
Cl ⁻ (mg/l)	92.3	284.0	211.58	234.3	83.01	1063
SAR	1.7	5.3	4.1	4.7	1.45	16
Pb (µg/l)	3.2	29.5	19.78	19.7	10.54	500
Cd (µg/l)	1.0	10.0	8.06	9.9	3.95	10
Zn (µg/l)	43.8	96.8	63.52	56.8	20.24	200
Mn (µg/l)	7.6	63.5	35.76	43.8	26.38	200
Cr (µg/l)	25.2	26.3	25.70	25.4	0.55	10
Ni (µg/l)	13.1	14.0	13.58	13.6	0.35	200
Cu (µg/l)	1.2	5.1	3.00	3.3	1.68	200
Fe (µg/l)	3.0	98.0	23.96	7.5	41.46	500
Co (µg/l)	10.1	11.4	10.80	11.0	0.61	5
Li (µg/l)	3.3	6.4	5.66	6.3	1.33	250

*FAO (1994)

⁻ Wide spread – heterogeneous distribution pattern; Cr, Co and Cu

Parameter	Ηq	EC	Ca	Mg	Na	K	HC03	D	SAR	Pb	G	Zn	Mn	с.	Ņ	ū	Fe	ට
Hd																		
EC	-0.64	-																
Ca	-0.79	0.70	_															
Mg	-0.71	0.98	0.79	-														
Na	-0.38	0.90	0.32	0.82	_													
K	-0.49	0.96	0.68	0.96	0.85	-												
HC03	-0.62	0.00	0.49	0.83	0.90	0.76	_											
G	-0.12	0.81	0.46	0.78	0.79	0.93	0.58											
SAR	0.01	0.60	-0.15	0.46	0.89	0.56	0.72	0.60	-									
Pb	0.80	-0.78	-0.84	-0.78	-0.54	-0.65	-0.80	-0.41	-0.18	-								
Cd	0.71	-0.71	-0.98	-0.77	-0.35	-0.68	-0.55	-0.50	0.08	06.0	-							
Zn	-0.49	-0.08	-0.13	-0.08	-0.01	-0.32	0.22	-0.60	0.02	-0.13	0.20	1						
Mn	-0.85	0.81	0.69	0.87	0.65	0.75	0.67	0.47	0.30	-0.60	-0.58	0.26	-					
Cr	0.42	-0.58	-0.22	-0.46	-0.64	-0.37	-0.86	-0.23	-0.61	0.71	0.35	-0.37	-0.26	1				
Ni	-0.29	-0.07	-0.13	0.02	0.00	-0.09	-0.11	-0.26	-0.03	0.32	0.34	0.55	0.48	0.35	1			
Cu	0.26	-0.81	-0.61	-0.83	-0.70	-0.94	-0.51	-0.96	-0.43	0.43	0.60	0.59	-0.60	0.08	0.12	1		
Fe	-0.44	0.10	-0.20	0.07	0.26	-0.10	0.35	-0.34	0.32	-0.03	0.30	0.91	0.42	-0.35	0.71	0.34	1	
C_0	0.18	-0.39	-0.51	-0.33	-0.21	-0.33	-0.43	-0.34	-0.05	0.71	0.68	0.32	0.09	0.58	0.89	0.25	0.51	1
11	ا ۱ کر	-0 06	1 13	00 U-	000-	-0 J	0.01	0 11	500	0.54	121	0.45	0 1K	0.5.0	40	0 V	010	0.0

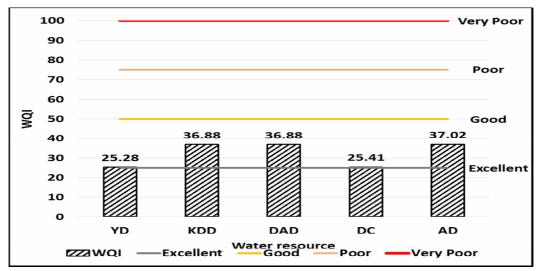
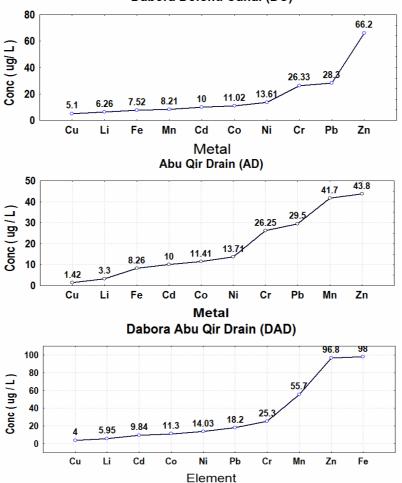


Fig. 2.WQI of Kafr El-Dawar water resources for irrigation utilizations



Dabora Defshu Canal (DC)

Fig 3. Increasing sequence of heavy metals of water resources DC, AD and DAD

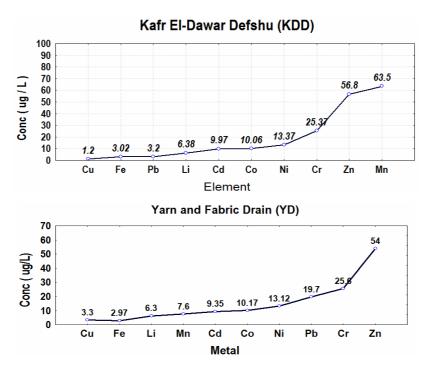


Fig 3 cont. Increasing sequence of heavy metals of water resources KDD and YD

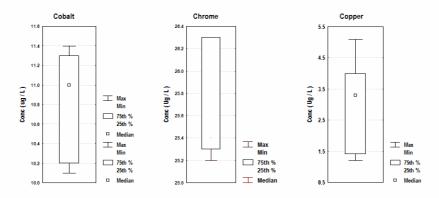


Fig 4, a. Wide spread - distribution pattern of Cr ,Co,Cu water resources

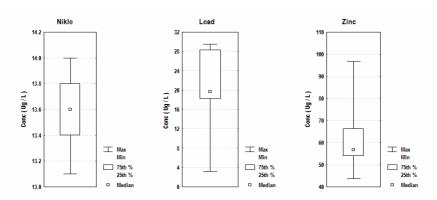


Fig 4. b.Distribution pattern : moderately spread of Pb , Ni and Zn in water resources

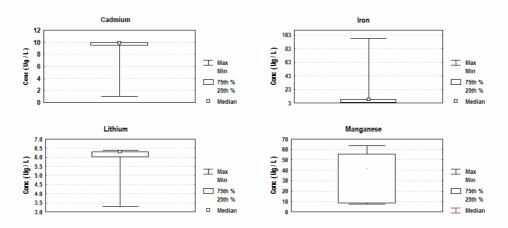


Fig 4.c. Distribution pattern : narrow spread of Fe, Cd, Li and Mn in water resources

Assessment of heavy metals pollution:

Heavy metals in water resources, in the cases of short and long term use, was assessed based on: (i) heavy metals conventional scale, (ii) indexing approach of contamination index (CI), and (iii) metallic Index (MI).

(i) Heavy Metals conventional scale:

A conventional scale was derived from recommended limits for constituents in reclaimed water for irrigation (EPA, 2004) to determine the classes of heavy metal in water pollution (Table 6). This conventional scale was based on MAC_s and MAC₁ that are maximum admissible concentrations (upper permissible limits) in cases of short and long term uses, respectively. The values of MAC_s and MAC₁ were considered as the lower limits of polluted class. The lower and upper limits of the safe concentrations of *i* th heavy metal assume to be less than 5 % and 10% of MAC_s and MAC₁. The risky class of water pollution by heavy metals was represented by a range from the value greater than the upper limit of safe class and the value of the lower limit of polluted class.

Table 6. Conventional interpretative scale of heavy metals in water

metal	_	Short Ter	m Use (µg/L)			Long Te	rm Use (µg/L)
metai	Ideal	Safety	Risky	Polluted	Ideal	Safety	Risky	Polluted
Cd	< 2.5	2.5-5.0	5-50	>50	< 0.5	0.5-1.0	1-10	>10
Co	< 250	250-500	500-5000	>5000	< 2.5	2.5-5	5-50	> 50
Cr	< 50	50-100	100-1000	>1000	< 5	5-10	10-100	> 100
Cu	< 250	250-500	500-5000	>5000	< 10	10-20	20-200	>200
Fe	< 1000	1000-2000	2000-20000	>20000	< 250	250-500	500-5000	> 5000
Li	< 125	125-250	250-2500	>2500	< 125	125-250	250-2500	>2500
Mn	< 500	500-1000	1000-10000	>10000	< 10	10-20	20-200	>200
Ni	< 100	100-200	200-2000	>2000	< 10	10-20	20-200	>200
Pb	< 500	500-1000	1000-10000	>10000	< 250	250-500	500-5000	>5000
Zn	< 500	500-1000	1000-10000	>10000	< 100	100-200	200-2000	>2000

Table 7. Classes of heavy metals pollution in case of short-term use, based on conventional interpretative scale of heavy metals pollution in water

Water					Clas	S				
resource	Cu	Fe	Li	Mn	Cd	Со	Ni	Pb	Cr	Zn
YD	Ideal	Ideal	Ideal	Ideal	Risky	Ideal	Ideal	Ideal	Ideal	Ideal
KDD	Ideal	Ideal	Ideal	Ideal	Risky	Ideal	Ideal	Ideal	Ideal	Ideal
DAD	Ideal	Ideal	Ideal	Ideal	Risky	Ideal	Ideal	Ideal	Ideal	Ideal
DC	Ideal	Ideal	Ideal	Ideal	Risky	Ideal	Ideal	Ideal	Ideal	Ideal
AD	Ideal	Ideal	Ideal	Ideal	Risky	Ideal	Ideal	Ideal	Ideal	Ideal

Water					Cla	ISS				
resource	Cu	Fe	Li	Mn	Cd	Со	Ni	Pb	Cr	Zn
YD	Ideal	Ideal	Ideal	Ideal	Risky	Polluted	Safe	Ideal	Risky	Ideal
KDD	Ideal	Ideal	Ideal	Polluted	Risky	Polluted	Safe	Safe	Risky	Ideal
DAD	Ideal	Safe	Ideal	Risky	Risky	Polluted	Safe	Ideal	Risky	Ideal
DC	Ideal	Ideal	Ideal	Ideal	Risky	Polluted	Safe	Ideal	Risky	Ideal
AD	Ideal	Ideal	Ideal	Risky	Risky	Polluted	Safe	Ideal	Risky	Ideal

 Table 8. Classes of heavy metals pollution in case of long-term use, based on conventional interpretative scale of heavy metals pollution in water

Table 9. Contamination factor (C_{fi}) and level of heavy metals in water

Water		C	ontamina	ition facto	or(C _{fi}) of	heavy me	etals in wa	ater		
resource	Cd	Со	Cr	Cu	Fe	Li	Mn	Ni	Pb	Zn
YD -	0.87	0.98	-0.74	-0.99	-1.00	-0.98	-0.99	-0.93	-0.98	-0.95
ID -	Risky	Risky	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe
KDD -	0.99	-0.98	-0.75	-1.00	-1.00	-1.00	-0.65	-0.93	-1.00	-0.95
KDD -	Risky	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe
DAD -	0.97	-0.98	-0.75	-1.00	-0.78	-0.98	-0.94	-0.93	-0.98	-0.90
DAD	Risky	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe
DC -	1.00	-0.98	-0.74	-0.99	-1.00	-0.98	-1.00	-0.94	-0.97	-0.93
DC	Risky	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe
AD -	1.00	-0.98	-0.74	-1.00	-1.00	-1.00	-0.96	-0.93	-0.97	-0.96
AD	Risky	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe	Safe

Application of the conventional interpretative scale on heavy metals water pollution indicated that the concentrations of heavy metals generally at the safe levels, at case of short term use, (Table 7). This had an exception represented by Cd concentration in all water resources which is risky. Also, classes of heavy metals pollution (case of long-term use) were assessed based on their conventional interpretative scale (Table 8). The data showed that Co concentrations of all studied water resources, occupied the polluted class. Meanwhile, Mn concentration of the water resource (KDD) represents polluted class at the long-term use. The risky pollution class was found in Cd and Cr in all water resources under investigation as well as Mn in DAD and AD resources.

Contamination index (CI)

The contamination index of the studied waters was calculted to assess indivdually the contamination degree of each heavy metal (Table, 9) .The data indicted that the values of contamination factor (C_{fi}) ranged from 0.0 to 1.00. The contamination factor of cadmium (C_{fCd}) had generally the highest values which refer that cadmium is the more effective polutant. This data showed that the majority of heavy metals had low values of contamination factor (C_{fi}), to point out that the studied water resources at safe use level. This certainly with expection of cadmium.

The contamination index (CI) was calculated in cases of short and long term use to assess the heavy metals pollution in water. For the cases of short use, the CI values ranged from -7.49 (water of AD) to 7.51 (water of DC). As for the cases of long-term use, the CI values extended from 5.49 (water of YD) to 33.20 (water of KDD). The contamination index was used as reference to evaluate the extent of metal pollution (Table 10). This interpretation scale of contamination was modified by introducing the case of long use. An additional development was introduced by considering the CI negative values to have safe water use. The data showed that the CI of water of (YD), (DAD) and (AD) had the negative values of -5.67, -7.26 and -7.49, respectively. This indicated that water in the studied resources are safe in use at the short time run. Contrary, all studied water resources were highly contaminated that they cannot be used at the long time run (Table, 11).

Table10.Interpretativescaleofcontamination index

Ν	Contaminatio	Contaminati	ion Index (CI)
	n Degree	Short Use	Long Use
1	Safe	< 0.0	< 0.0
2	Low	$0.0 - \le 1$	$0.0 - \le 0.5$
3	Medium	>1 - < 3	> 0.5 - < 1.5
4	High	\geq 3	≥ 1.5

Water		Contamination	Index (CI)	
resource	Short	-term Use	Long	-term Use
	value	Class	value	Class
YD	-5.67	Safe	5.49	High
KDD	2.80	Medium	33.20	High
DAD	-7.26	Safe	9.75	High
DC	7.51	High	6.61	High
AD	-7.49	Safe	8.06	High

Table 11.	Contaminatio	on Index of studie	d water resources
1 and 11.	Contamination	III IIIUCA UI SLUUIC	u matti i tovui tto

Metal index (MI):

Another index is used to estimate the pollution of water by heavy metals under investigation for irrigation utilization. Metal index denotes the trend evaluation of the present status by computing all measured metals (Table 12). According to the values of metal index, four of five selected drains or canal are seriously threatened with metal pollution for irrigation usage (MI>1), since MI reaches to 9.12 at DAD drain for irrigation usages. Only KDD drain has low value of MI (-0.21) less than MI threshold of warning value. Therefore, no heavy metal problems expected as a result of use KDD drain for irrigation usage.

Table 12. Metal index in Kafr El-Dawarwaterresourcesforirrigationwaterutilizations

Water Resource	MI value	MI class
YD	5.50	Threshold of warning
KDD	-0.21	No warning
DAD	9.12	Threshold of warning
DC	6.60	Threshold of warning
AD	8.06	Threshold of warning

Conclusion and recommendation

Water resources in Kafr El-Dawar region receive wastewater from textile companies which suppress the water quality. Although WQI results showed that the water quality of the studied resources is good for irrigation usage according to the selected parameters in the present study. Using contamination index to evaluate heavy metal pollution indicated that water in these resources are safe in use at the short time run. However, high contaminated was noticed at the long time run. Also, metal index parameter agree with the data of contamination index except in water of KDD source. Therefore, the study recommends tightening the control on the discharged wastewaters into these water resources, to meet with the effluent concentration discharge standards set in Egyptian Law 48/1982 for the protection of the waterways in the region against pollution.

REFERENCES

- Al-Mohammed, F.M. and A.A. Mutasher. 2013. Application of water quality index for evaluation of groundwater quality for drinking purpose in Dibdiba Aquifer, Kerbala City, Iraq. J. Babylon Univ./Eng. Sci. 21 (5): 1647–1660.
- American Public Health Association. 2005. Standard Methods for the Examinations of Waters and Waste Waters, 21st ed. APHA AWWA-WEF, Washington DC.
- Ayers R.S. and D.W. Westcot. 1994. Food, Agriculture Organization of the United Nations (FAO), Water Quality for Agriculture, Irrigation and Drainage, Rome, Paper No. 29. Rev. 1, M-56. ISBN 92-5-102263-1.
- Bakan, G., H. Boke Ozkoc, S. Tulek and H. Cuce. 2010. Integrated environmental quality assessment of Kızılırmak River and its coastal environment. Turk. J. Fisheries Aquat. Sci. 10: 453–462.
- Balan, I.N., M. Shivakumar and P.D.M. Kumar. 2012. An assessment of ground water quality using water quality index in Chennai Tamil Nadu, India. Chronicles Young Scient. 3 (2): 146–150.
- Brraich, O.S. and S. Jangu. 2015 Evaluation of Water Quality Pollution Indices for Heavy Metal Contamination Monitoring in the Water of Harike Wetland (Ramsar Site) India International Journal of Scientific and Research Publications, 5 (2): 2250-3153. Available @ http://www.ijsrp.org/research-paper-0215/ijsrpp38104.pdf
- Chowdhury, R.M., S.Y. Muntasir and M.M. Hossain. 2012. Water quality index of water bodies along Faridpur-Barisal road in Bangladesh. Glob. Eng. Tech. Rev. 2 (3): 1–8.
- Elshemy, M. and G. Meon. 2011. Climate Change Impacts On Water Quality Indices of the Southern Part of Aswan High Dam Reservoir, Lake Nubia. Fifteenth International Water Technology Conference, IWTC-15, Alexandria, Egypt, p. 17.
- EPA (Environment Protection Agency). 2004. Guidelines for Water Reuse, EPA- USA /625/R-04/108; Table 2-13, pages 167-170. Available @ http://www.lacsd.org/civica/filebank/ blobdload.asp?BlobID=2184
- Hakanson, L. 1980. An ecological risk index for aquatic pollution control, a sedimentological approach. Water Res. 14: 975–1001.

- Ibrahim, A.T.A. and H.M. Omar. 2013. Seasonal variation of heavy metals accumulation in muscles of the African Catfish Clarias gariepinus, River Nile water and sediments at Assiut Governorate, Egypt. J. Biol. Earth Sci. 3 (2): 236–248.
- Karbassi, A. and G.R.N. Bayati. 2005 Environmental geochemistry of heavy metals in a sediment core off Bushehr, Persian Gulf. Iran. J. Environ. Heal. Sci. Eng. 2: 255-260.
- Khalil, M.K.H., A.M. Radwan and K.H.M. El-Moselhy. 2007. Distribution of phosphorus fractions and some of heavy metals in surface sediments of Burullus lagoon and adjacent Mediterranean Sea. Egypt. J. Aquat. Res. 33 (1): 277–289.
- Lumb, A., T.C. Sharma and J.F. Bibeault. 2011. A Review of Genesis and Evolution of Water Quality Index (WQI) and Some Future Directions. Water Qual. Expo Health 3:11–24.
- Goher, M.E., A.M. Hassan, I.A. Abdel-Moniem, A.H. Fahmy and S.M. El-sayed. 2014. Evaluation of surface water quality and heavy metal indices of Ismailia Canal, Nile River, Egypt. Egy. J. Aquatic Res. 40: 225–233.
- Manoj, K., P.K. Padhy and S. Chaudhury. 2012 Study of Heavy Metal Contamination of the River Water through Index Analysis Approach and Environmetrics. Bull. Environ. Pharmacol. Life Sci. 1 (10): 07-15. Available @ www.bepls.com
- Mohrir A., D.S. Ramteke, C.A. Moghe, S.R. Wate and R. Sarin. 2002. Surface and groundwater quality assessment in Bina region. I.J.E.P. 22 (9): 961-969.

- Mwegoha, W.J.S. and C. Kihampa. 2010 Heavy metal contamination in agricultural soils and water in Dar es Salaam city, Tanzania. African J. Environ. Sci. Techno. 4 (11): 763-769. Available @ http://www.academicjournals.org/AJEST
- Page, A.L., H. Miller and D.R. Keeney. 1982. Methods of Soil Analysis. Part 2, ASA, SSSA, Madison, Wisconsin USA.
- Rumsey, D.J. 2016 What a Boxplot Can Tell You about a Statistical Data Set available @ http://www.dummies.com/how-to/content/what-a-boxplot-can-tell-you-about-a-statistical-da.html.
- Sany, S.B., A. Salleh., M. Rezayi., N. Saadati and L. Narimany. 2013 Distribution and Contamination of Heavy Metal in the Coastal Sediments of Port Klang, Selangor, Malaysia. Water Air Soil Pol. 224:1476-1486.
- Sasmaz, A., E. Obek and H. Hasar. 2008. The accumulation of heavy metals in *Typha latifolia* L. grown in a stream carrying secondary effluent. Ecological Engineering. 33: 278-284.
- Tamasi, G. and R. Cini. 2004. Heavy metals in drinking waters from Mount Amiata. Possible risks from arsenic for public health in the province of Siena. Sci. Total Environ. 327: 41–51.
- Tripaty, J.K. and K.C. Sahu. 2005. Seasonal hydrochemistry of groundwater in the Barrier Spit system of the Chilika Lagoon, India. J.Environ. Hydrol. 13: 1–9.
- Tyagi1, S., B. Sharma, P. Singh and R. Dobhal. 2013. Water Quality Assessment in Terms of water quality index. Am. J. Water Res. 1(3): 34–38.

_

